

This Page Is Inserted by IFW Operations  
and is not a part of the Official Record

## **BEST AVAILABLE IMAGES**

Defective images within this document are accurate representations of the original documents submitted by the applicant.

Defects in the images may include (but are not limited to):

- BLACK BORDERS
- TEXT CUT OFF AT TOP, BOTTOM OR SIDES
- FADED TEXT
- ILLEGIBLE TEXT
- SKEWED/SLANTED IMAGES
- COLORED PHOTOS
- BLACK OR VERY BLACK AND WHITE DARK PHOTOS
- GRAY SCALE DOCUMENTS

**IMAGES ARE BEST AVAILABLE COPY.**

**As rescanning documents *will not* correct images,  
please do not report the images to the  
Image Problem Mailbox.**



















# THE MERCK INDEX

AN ENCYCLOPEDIA OF  
CHEMICALS, DRUGS, AND BIOLOGICALS

ELEVENTH EDITION

Susan Budavari, *Editor*  
Maryadele J. O'Neil, *Associate Editor*  
Ann Smith, *Assistant Editor*  
Patricia E. Heckelman, *Editorial Assistant*

*Published by*  
**MERCK & CO., INC.**  
RAHWAY, N. J., U. S. A.

1989

Copyright © 1989 by Merck & Co., Inc.  
Previous Editions  
Copyright © 1940, 1952, 1960, 1968, 1976, 1983  
by Merck & Co., Inc.  
All rights reserved under the international copyright  
conventions. Copyright under the Universal Copyright  
Convention.

The Merck Index is published on a non-profit basis as a  
service to the scientific community.

**Merck & Co., Inc.**

Rahway, New Jersey, U.S.A.

MERCK SHARP & DOHME  
West Point, Pa.

MERCK SHARP & DOHME INTERNATIONAL  
Rahway, N.J.

MERCK SHARP & DOHME RESEARCH LABORATORIES  
Rahway, N.J./West Point, Pa.

MSD AGVET DIVISION  
Woodbridge, N.J.

HUBBARD FARMS, INC.  
Walpole, N.H.

MERCK CHEMICAL MANUFACTURING DIVISION  
Woodbridge N.J.

MERCK PHARMACEUTICAL MANUFACTURING DIVISION  
Rahway, N.J.

CALGON CORPORATION  
*Water Management Division*  
Pittsburgh, Pa.  
*Calgon Vestal Laboratories*  
St. Louis, Mo.

KELCO DIVISON  
San Diego, Ca.

---

1st Edition—1889  
2nd Edition—1896  
3rd Edition—1907  
4th Edition—1930  
5th Edition—1940  
6th Edition—1952  
7th Edition—1960  
8th Edition—1968  
9th Edition—1976  
10th Edition—1983  
11th Edition—1989

---

Library of Congress Catalog  
Card Number 89-60001  
ISBN Number 911910-28-X

Printed in the U.S.A  
First Printing—November 1989  
Second Printing—February 1990  
Third Printing—September 1991

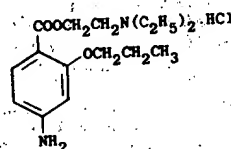
# Propoxyphene

7851

Minute crystals, mp 91.5°. Dec at high temp forming methyl isocyanate. Sol in methanol, acetone and many organic solvents, but only slightly sol in cold hydrocarbons. Water soly about 0.2% at 20°. Unstable in highly alkaline media. LD<sub>50</sub> orally in male, female rats: 83, 86 mg/kg (Gaines).

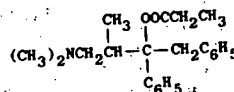
USE: Insecticide.

**7850. Propoxycaine Hydrochloride.** 4-Amino-2-propoxybenzoic acid 2-diethylaminoethyl ester hydrochloride; 2-diethylaminoethyl 4-amino-2-propoxybenzoate hydrochloride; 2-diethylaminoethyl 2-propoxy-4-aminobenzoate hydrochloride; Ravocaine hydrochloride; Pravocaine hydrochloride; Blockaine hydrochloride. C<sub>16</sub>H<sub>21</sub>ClN<sub>2</sub>O<sub>3</sub>; mol wt 330.86. C 58.08%, H 8.23%, Cl 10.72%, N 8.47%, O 14.51%. Prepn: Clinton, Laskowski, U.S. pat. 2,689,248 (1954 to Sterling Drug).

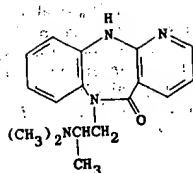


White, odorless crystals, mp 148-150°. Discolors upon prolonged exposure to light and to air. Freely sol in water; sol in ethanol, chloroform. Sparingly sol in ether. Practically insol in acetone, chloroform. pH of a 2% aq soln 5.4. THERAP CAT: Local anesthetic.

**7851. Propoxyphene.** [S-(R\*,S\*)]-α-[2-(Dimethylamino)-1-methylethyl]-α-phenylbenzeneethanol propanoate (ester); α-d-4-dimethylamino-3-methyl-1,2-diphenyl-2-butanol propanoate; (+)-1,2-diphenyl-2-propionyloxy-3-methyl-4-dimethylaminobutane; (+)-4-dimethylamino-1,2-diphenyl-3-methyl-2-propionyloxybutane; d-propoxyphene; dextro-3-methyl-2-propionyloxybutane; d-propoxyphene; C<sub>21</sub>H<sub>25</sub>NO<sub>2</sub>; mol wt 339.48. C 77.83%, H 8.61%, N 4.13%, O 9.43%. Prepn of racemate: Pohland, Sullivan, J. Am. Chem. Soc. 75, 4458 (1953); Pohland, U.S. pat. 2,728,779 (1955 to Lilly). Prepn of (+)-form: Pohland, Sullivan, J. Am. Chem. Soc. 77, 3400 (1955). Stereochemistry: Sullivan et al., J. Org. Chem. 28, 2381 (1963); Casey, Myers, J. Pharm. Pharmacol. 16, 455 (1964). Stereocenter synthesis: Pohland et al., J. Org. Chem. 28, 2483 (1963). Metabolism: S. L. Due et al., Biomed. Mass Spectrom. 3, 217 (1976). The α-dl- and d-diastereoisomers possess marked analgesic activity in contrast to the β-diastereoisomers which are substantially inactive. Toxicity: E. I. Goldenthal, Toxicol. Appl. Pharmacol. 18, 185 (1971); J. L. Emerson et al., ibid. 19, 445 (1971). Comprehensive description: B. McEwan in Analytical Profiles of Drug Substances vol. 1, K. Florey, Ed. (Academic Press, New York, 1972) pp 301-318. Symposium on pharmacology, toxicology, and clinical efficacy of propoxyphene alone and in combination with acetaminophen: Human. Toxicol. 3, Suppl., 1S-238S (1984).

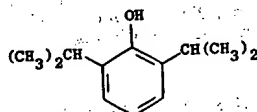


Crystals from petr ether, mp 75-76°. [α]<sub>D</sub><sup>25</sup> +67.3° (c = 0.6 in chloroform). α-d-Hydrochloride, C<sub>21</sub>H<sub>25</sub>ClNO<sub>2</sub>, Algan, Antalvic, Darvon, Depromic, Depranol, Develin, Dolene, Dolocap, Doraphen, Erantin, Femadol, Harmar, Propox, Propoxychel, Proxagesic. Bitter crystals from methanol + ethyl acetate, mp 163-168.5°. [α]<sub>D</sub><sup>25</sup> +59.8° (c = 0.6 in water). Sol in water, alc, chloroform, acetone. Practically insol in benzene, ether. LD<sub>50</sub> in mice, rats (mg/kg): 28, 15 i.v.; 111, 58 i.p.; 211, 134 s.c.; 282, 230 orally (Emerson). α-d-Form napsylate monohydrate, C<sub>22</sub>H<sub>27</sub>NO<sub>5</sub>·H<sub>2</sub>O, Darvon-N, Doloxene. LD<sub>50</sub> orally in female rats: 990 mg/kg (Goldenthal). α-l-Form, see Levopropoxyphene.



mp 122°. Hydrochloride, C<sub>17</sub>H<sub>21</sub>ClN<sub>2</sub>O, UP 106, Depressin, Vagran. THERAP CAT: Antidepressant.

**7847. Propofol.** 2,6-Bis(1-methylethyl)phenol; 2,6-diisopropylphenol; disoprofol; ICI 35868; Diprivan; Disoprivan; Rappanovet. C<sub>12</sub>H<sub>18</sub>O; mol wt 178.27. C 80.85%, H 10.18%, O 8.97%. Prepn: A. J. Kolka et al., J. Org. Chem. 21, 712 (1956); 22, 642 (1957); G. G. Ecke, A. J. Kolka, U.S. pat. 2,831,898 (1958 to Ethyl Corp.); T. J. Kealy, D. D. Coff, J. Org. Chem. 26, 987 (1961); B. E. Firth, T. J. Rosen, U.S. pat. 4,447,657 (1984 to Universal Oil Products). Chromatographic study: J. K. Carlton, W. C. Bradbury, J. Am. Chem. Soc. 78, 1069 (1956). Animal studies: J. B. Allen, Brit. J. Anaesth. 52, 731 (1980). Pharmacokinetics: H. K. Adam et al., ibid. 743; idem, ibid. 55, 97 (1983). Determination in blood: idem, J. Chromatog. 223, 232 (1981). Comparative studies vs other injectable anesthetics: B. Kay, D. K. Stephenson, Anaesthesia 35, 1182 (1980); D. V. Rutter et al., ibid. 1188. Use in i.v. anesthesia: E. Major et al., ibid. 37, 541 (1982). Cardiovascular effects: D. Al-Khudhairi et al., ibid. 1007. Pharmacology of emulsion formulation: J. B. Glen, S. C. Hunter, Brit. J. Anaesth. 56, 617 (1984). Series of articles on pharmacology and clinical experience: Postgrad. Med. J. 61, Suppl. 3, 1-169 (1985).

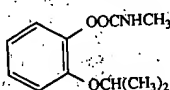


mp 136°. bp<sub>17</sub> 126°. mp 19°. n<sub>D</sub><sup>20</sup> 1.5134. n<sub>D</sub><sup>25</sup> 1.5111. d<sub>20</sub> 0.955.

THERAP CAT: Anesthetic (intravenous). THERAP CAT (VET): Intravenous anesthetic (dogs and cats).

**7848. Propolis.** Bee bread; hive dross. A resinous substance found in beehives. Collected by bees from buds. Rich in caffeic acid from propolis: Cizmárik, Matel, Experientia 26, 713 (1970). Antimicrobial constituents of propolis: J. Metzner et al., Pharmazie 30, 799 (1975); E. M. Schneidewind et al., ibid. 803. Review on the origin, chemical constituents and therapeutic activity: M. H. Haydak, State of Iowa Repts. State Apiarist 1953, p 74-87; M. Vanhaelen, Vanhaelen-Fastre, J. Pharm. Belg 34, 253 (1979). Greenish-brown, sticky mass. Aromatic odor. d 1.2. mp 64°. Becomes brittle when cooled below 15°. Extraction with alcohol gives propolis wax. The residue from the alcohol extraction is called propolis resin, yielding propolis balsam on extraction with hot petr ether. Propolis balsam has a hyacinth odor and is said to contain 10% cinnamyl alcohol.

**7849. Propoxur.** 2-(1-Methylethoxy)phenol methylcarbamate; o-isopropoxyphenyl N-methylcarbamate; aprocarb; BAY 39007; BAY 9010; Baygon; Bifex; Blattanax; Invisi-Gard; Propyon; Suncide; Sendran; Uden. C<sub>11</sub>H<sub>15</sub>NO<sub>3</sub>; mol wt 209.24. C 63.14%, H 7.23%, N 6.69%, O 22.94%. Prepn: U.S. pat. 3,111,539 (1963 to Bayer; Chemagro Corp.). Properties: Pflanzenschutz Nachr. Bayer 18, 53 (1965). Toxicity data: T. B. Gaines, Toxicol. Appl. Pharmacol. 14, 151 (1969). Teratogenicity study: K. D. Courtney et al., J. Environ. Sci. Health B20, 373 (1985).



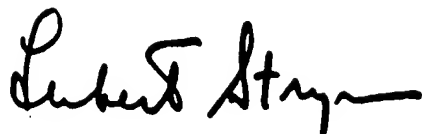
base bp<sub>3</sub> 140-144°. n alcohol, ether. A

lamino)propyl]-1,6-izepin-5-one; 6,11-ylethyl]-5H-pyrido-H<sub>2</sub>N<sub>2</sub>O; mol wt 209.24. C 63.14%, H 7.23%, N 6.69%, O 22.94%. Prepn: bs. U.P.S.A.; Hoff-966, 2316. Psycho-6, 451 (1971).

Consult the cross index before using this section.

# BIOCHEMISTRY

THIRD EDITION



LUBERT STRYER

STANFORD UNIVERSITY



W. H. FREEMAN AND COMPANY / NEW YORK

Library of Congress Cataloging-in-Publication Data

Stryer, Lubert.  
Biochemistry.

Includes index.

I. Biochemistry. I. Title.

QP514.2.S66 1988 574.19'2 87-36486

ISBN 0-7167-1843-X

ISBN 0-7167-1920-7 (international student ed.)

Copyright © 1975, 1981, 1988 by Lubert Stryer

No part of this book may be reproduced by any mechanical, photographic, or electronic process, or in the form of a phonographic recording, nor may it be stored in a retrieval system, transmitted, or otherwise copied for public or private use, without written permission from the publisher.

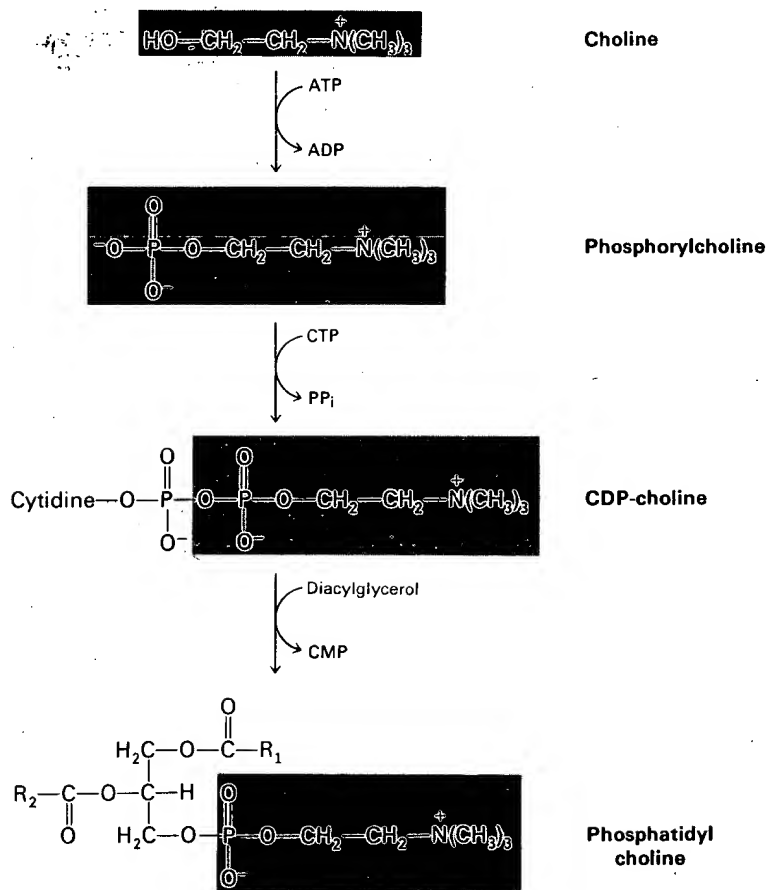
Printed in the United States of America

6 7 8 9 0 RRD 9 9 8 7 6 5 4 3 2 1 0

**PHOSPHOGLYCERIDES CAN ALSO BE SYNTHESIZED FROM A CDP-ALCOHOL INTERMEDIATE**

In mammals, phosphatidyl choline is synthesized by a pathway that utilizes choline obtained from the diet (Figure 23-2). Choline is phosphorylated by ATP to *phosphorylcholine*, which then reacts with CTP to form *CDP-choline*. The phosphorylcholine unit of CDP-choline is then transferred to a diacylglycerol to form *phosphatidyl choline*. Note that the activated species in this pathway is the cytidine derivative of phosphorylcholine rather than of phosphatidate.

**Figure 23-2**  
Synthesis of phosphatidyl choline.



Likewise, *phosphatidyl ethanolamine* can be synthesized from ethanolamine by forming a CDP-ethanolamine intermediate by analogous reactions. Alternatively, phosphatidyl ethanolamine can be formed from phosphatidyl serine by the enzyme-catalyzed exchange of ethanolamine for the serine moiety of the phospholipid.

**PLASMALOGENS AND OTHER ETHER PHOSPHOLIPIDS ARE FORMED FROM DIHYDROXYACETONE PHOSPHATE**

Some phospholipids contain an ether unit instead of an acyl unit at  $\text{C}_1$ . *Glycerol ether phospholipids* are synthesized starting with dihydroxyacetone phosphate (Figure 23-3). Acylation by a fatty acyl CoA yields a 1-acyl derivative that exchanges with a long-chain alcohol to form an ether at  $\text{C}_1$ . The keto group at  $\text{C}_2$  is reduced by NADPH, and the

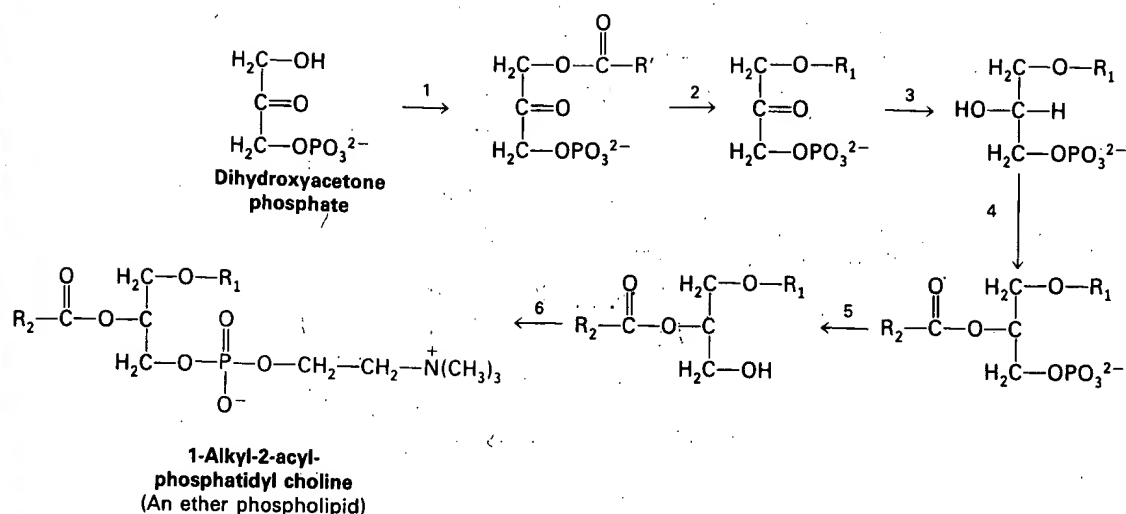
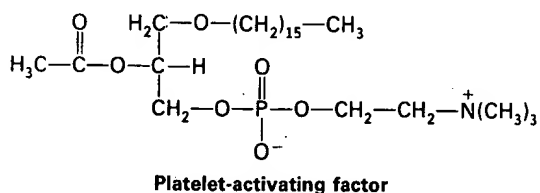


Figure 23-3

Synthesis of an ether phospholipid. The steps are (1) acylation by fatty acyl CoA, (2) exchange of an alcohol for the carboxylate moiety, (3) reduction by NADPH, (4) acylation by a second fatty acyl CoA, (5) hydrolysis of the phosphate ester, (6) transfer of a phosphocholine moiety.

resulting alcohol is acylated by a long-chain CoA. Removal of the 3-phosphate group yields 1-alkyl-2-acylglycerol, which reacts with CDP-choline to form the ether analog of phosphatidyl choline.

An ether phospholipid with striking activities has recently been identified. *Platelet-activating factor* is a 1-alkyl-2-acetyl ether analog of phosphatidyl choline. Even a very low concentration of this compound (0.1 nM) in the blood causes the aggregation of platelets and the dilation of blood vessels. The presence of an acetyl group rather than a long-chain acyl group at C<sub>2</sub> increases the water-solubility of this lipid, enabling it to function in an aqueous environment.



*Plasmalogens* are phospholipids containing an  $\alpha,\beta$ -unsaturated ether at C<sub>1</sub>. Phosphatidal choline, the plasmalogen corresponding to phosphatidyl choline, is formed by desaturation of a 1-alkyl precursor. The desaturase catalyzing this final step in the synthesis of a plasmalogen is a microsomal enzyme akin to the one that introduces double bonds into long-chain fatty acyl CoAs: O<sub>2</sub> and NADH are reactants, and cytochrome b<sub>5</sub> participates in catalysis (p. 489).

